

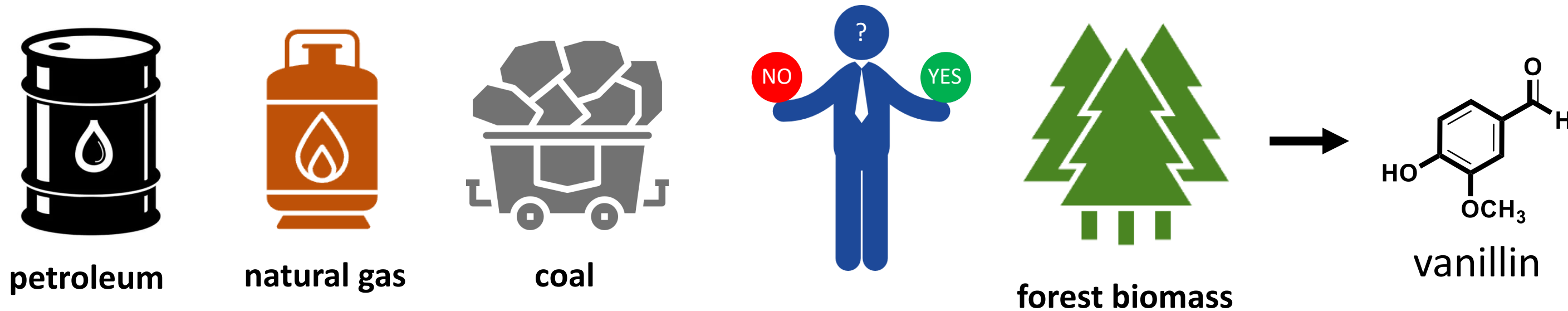
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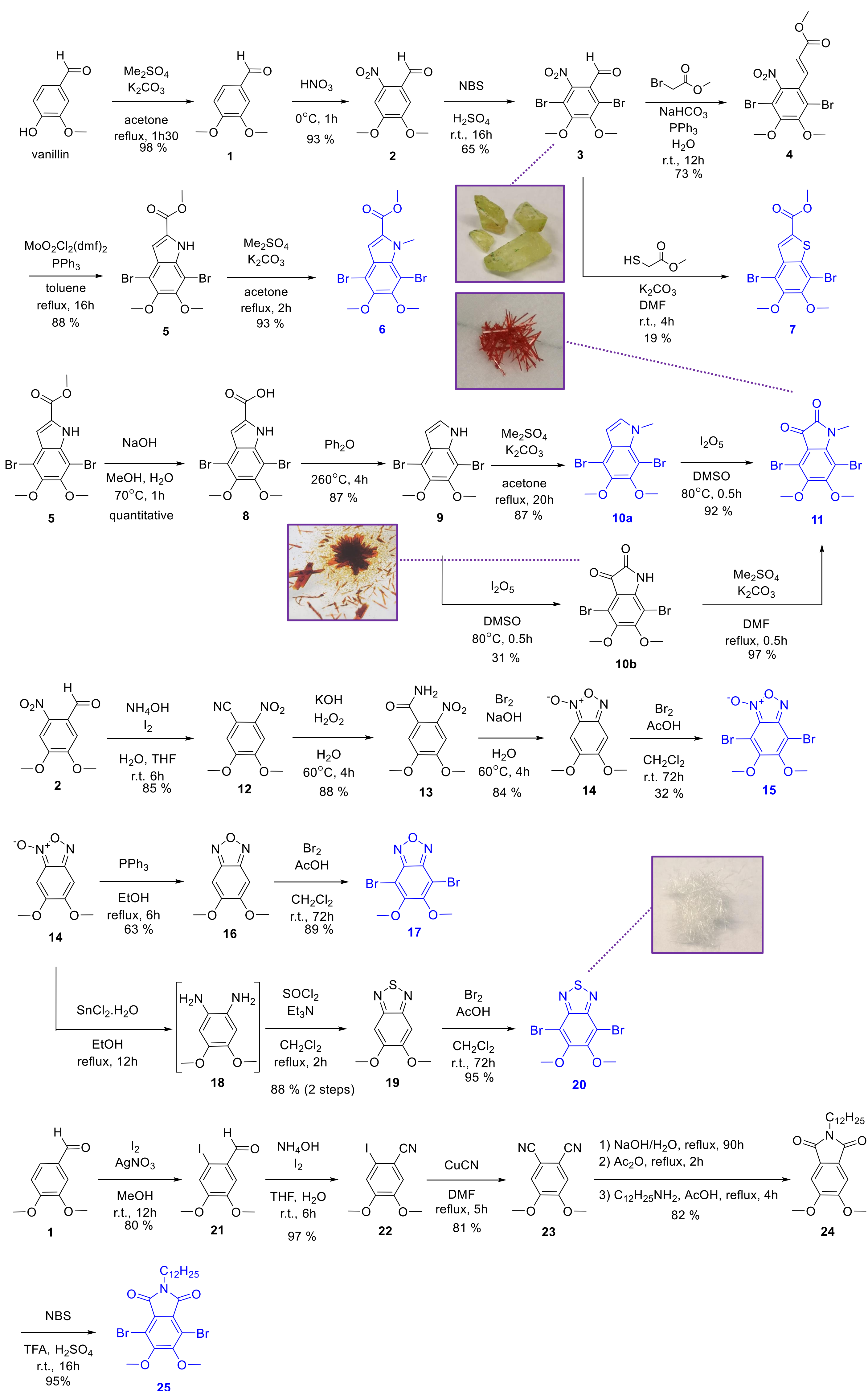
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Introduction

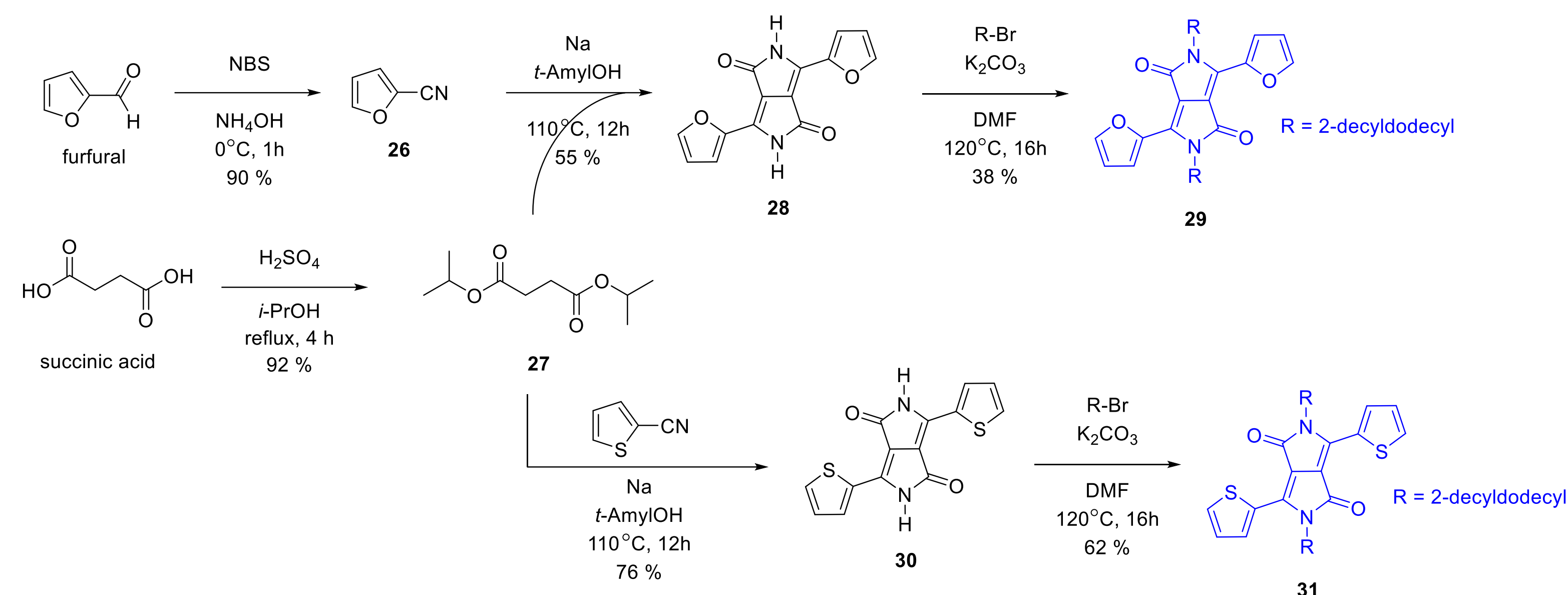
Electronic is so well integrated into our lives that it would be difficult to imagine a single day without the support of modern technology. Undeniably, these electronic products (cellphones, tablets, portable computers) have become essential tools in our daily routine. However, the fact remains that their lifespan is limited and that the resources used for their manufacture are demanding on the environment. A drastic change in the way we harness resources and manage electronic waste is required to minimize negative impacts on our environment. In order to eliminate electronic waste, materials from renewable and recyclable sources are sought. In this regard, forest biomass is considered the only sustainable source of organic carbon and therefore the ideal replacement of petroleum products to produce sustainable chemical compounds. One of the resources derived from biomass, lignocellulose, is the most abundant bio-based material on earth. Indeed, the main value-added compound obtained from the depolymerization of lignin is vanillin. Vanillin is an aromatic compound with three functional groups which can be chemically modified (methoxy, aldehyde and hydroxyl). The present work describes how the vanillin moiety can be expanded into several novel building blocks for the preparation of sustainable conjugated polymeric materials.



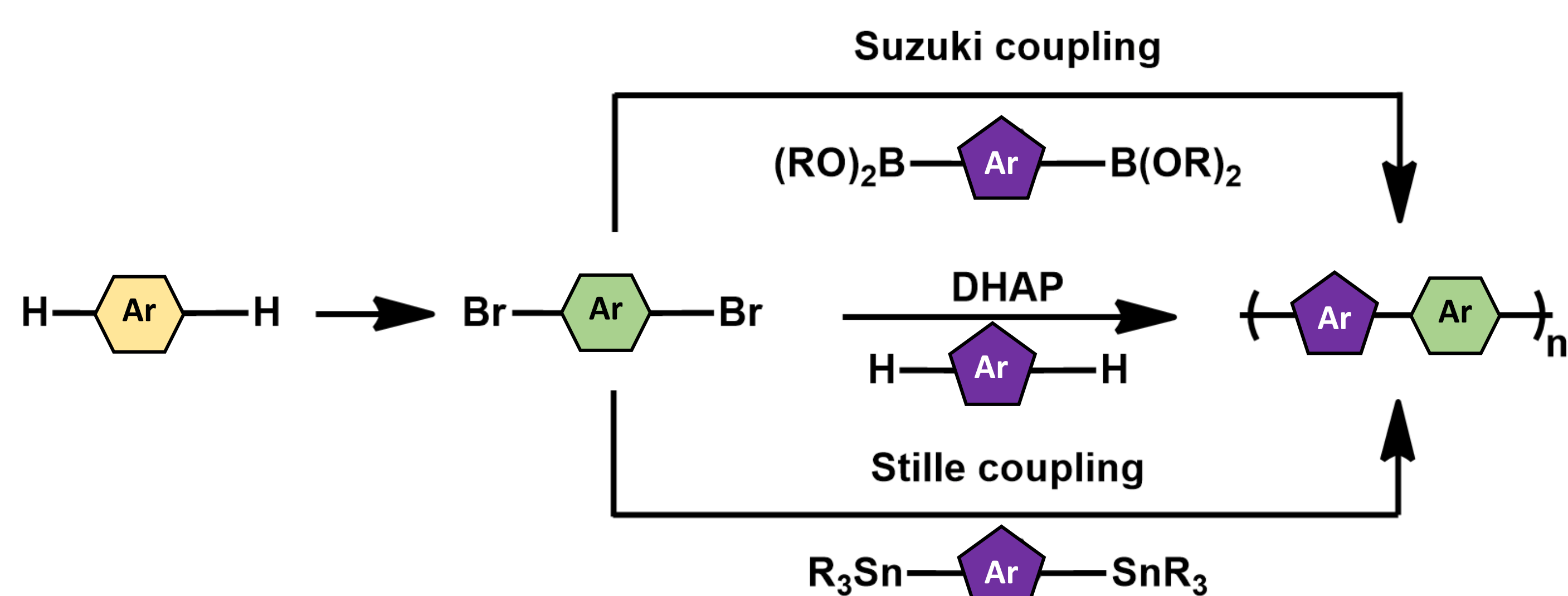
Vanillin-based monomers synthesis



DPP furan-based monomers synthesis



Polymerization



STILLE COUPLING

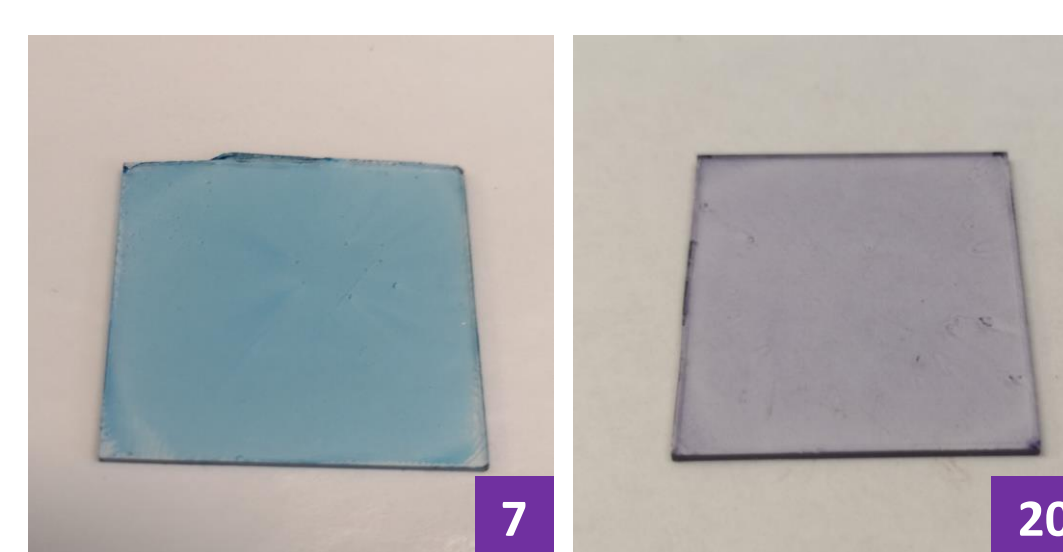
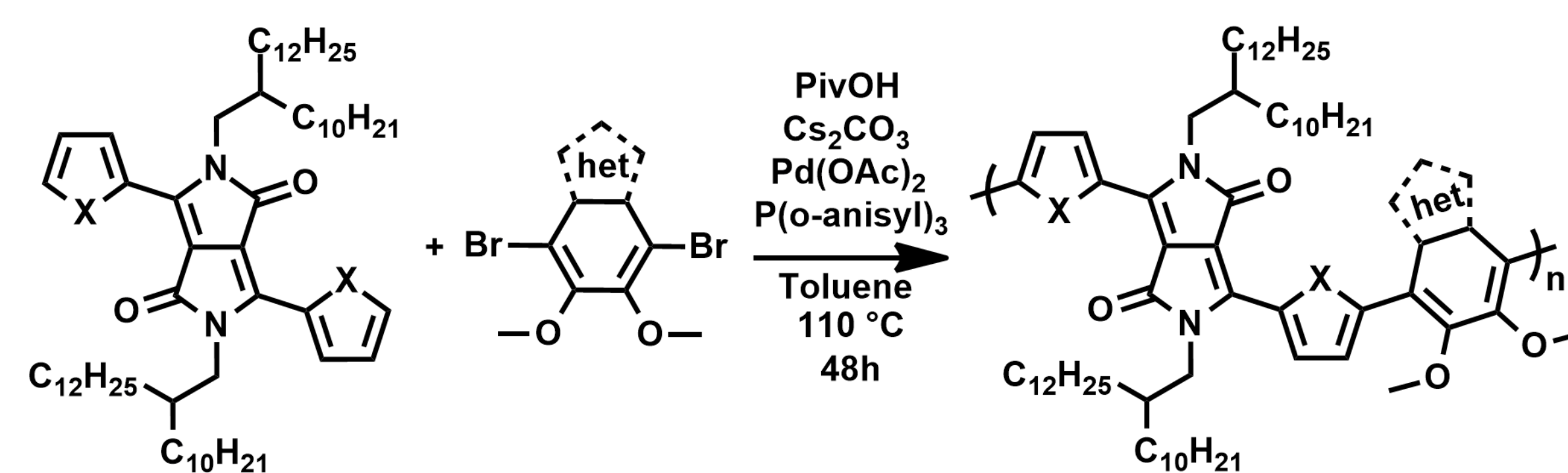
- More synthetic steps
- Highly toxic reactants
- Highly toxic byproducts

SUZUKI COUPLING

- Adaptable to aqueous conditions
- More synthetic steps
- Unstable and/or more costly reactants

DHAP

- Adaptable to aqueous conditions
- Less synthetic steps
- No highly toxic byproducts
- Less costly reactants



Biosourced monomer	DPP*	E _g (eV)	HOMO (eV)	LUMO (eV)	M _n (kDa)	M _w (kDa)
7	Th	1.59	-5.29	-3.75	17	34
6	Th	1.72	-5.44	-3.68	19	45
25	Fu	1.67	-5.01	-3.24	6	12
	Th	1.62	-5.31	-3.60	10	18
11	Fu	1.83	-5.12	-3.23	10	22
	Th	-	-	-	-	-
20	Th	1.31	-5.36	-3.73	2	3
17	Th	1.36	-5.41	-3.79	1	2
15	Th	-	-	-	-	-

*Th= thiophene, Fu= furane

References

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- Sonar, P.; Singh, S.P.; Williams, E.L.; Li, Y.; Soh, M.S.; Dodabalapur, A.; Furan containing diketopyrrolopyrrole copolymers: synthesis, characterization, organic field effect transistor performance and photovoltaic properties, *J. Mater. Chem.* **2011**, *22*, 4425-4435.